

| Course name                                   | General Chemistry with The Elements of Physics   |
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| Entity running the course                     | Faculty of Ceramics and Glass / Department of Conservation and Restoration of Ceramics and Glass   |
| Entity for which the course has been prepared |  |
| Course type                                   | Core course, compulsory course.  |
| Year of study/semester;<br>Type of studies    | Year II, semester 3 and 4, full time master's degree studies   |
| ECTS credits                                  | Semester 3: 2 ECTS credits<br>Semester 4: 1 ECTS credits   |
| Academic tutor                                | D.Sc. Zbigniew Burski  |
| Aim of the course                             | The aim of the course is to make the student understand the use of inorganic and organic compounds in the conservation and restoration of glass and ceramics.  |
| Prerequisites                                 | The I year of the study passed.  |
| Learning outcomes:                            |  |
| – knowledge                                   | The student has an extended knowledge of the physico-chemical phenomena occurring in glass and ceramics.  They know the basic concepts and laws of chemistry.  They know the properties of mineral binders.  They know what the organic compounds properties are.  They know the most important properties of plastics, stone, glass and ceramics, including corrosion processes.  |
| – skills                                      | The student can apply their knowledge in analyzing the corrosion phenomena in glass and ceramics.  They can compare the properties of organic and inorganic compounds.  They can analyze basic chemical reactions. They are able to point out corrosive destruction in plastics, stone, glass and ceramics.  |
| – personal and social competence              | The student initiates work in a team. They put the acquired knowledge into practice. They understand the need for extending their knowledge.   |
| Course content                                | Fundamental concepts and laws of chemistry. Chemical reactions. Atomic structure. Chemical bonds. The stoichiometry of chemical equations. States of matter concentration - gas, liquid /dissociation, pH, hydrolysis, hydration, solubility product/, solid /crystallographic systems, polymorphism, isomorphism, actual crystals /. Physical and chemical properties of mineral binders. Chemistry of the organics /aliphatic or aromatic/. Polymers. The properties and corrosion of plastics, stone, glass and ceramics. |
| Course form and number of course hours        | Lectures - 30 hours/sem.   |
| Assessment methods and criteria               | Semester 3: 100% active participation in classes Semester 4: 25% active participation in classes, 75 % pass – in a writing form.   |
| Assessment type                               | Semester 3: pass.<br>Semestr 4: graded pass  |
| Literature                                    | 1. I.Barycka, K.Skudlarski, "Podstawy chemii" /Fundamentals of chemistry/ 2. L.Jones, P. Atkins, "Chemia ogólna" /General chemistry/ 3. L.Czarnecki i inni, "Chemia w budownictwie" /Chemistry in architecture/  |
| Teaching aids                                 |  |
| Language of instruction                       | Polish   |
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